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2 Measurements of suprathermal hydrogen flux in CASTOR tokamak

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8 Abstract

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9 A study of the non-equilibrium absorption of hydrogen particles by a plasma-facing absorption probe (AP) made of Group V metals in 10 the CASTOR tokamak environment has been undertaken with the main purposes: (1) to investigate the role of nonmetallic coatings upon 11 plasma-facing materials in D/T inventory and recycling and (2) to develop a method of the registration and diagnostic of the flux of supra-12 thermal hydrogen. Two series of experiments were performed: with AP of Nb and of V. The absorption of suprathermal hydrogen particles 13 coming from the tokamak plasma was investigated as a function of AP temperature, AP distance from the plasma, AP bias and of plasma 14 pulse duration. The possibility of a reliable registration of suprathermal hydrogen was demonstrated in spite of the short plasma pulse 15 duration and a relatively high background H₂ pressure. The composition of the hydrogen flux (molecules, atoms, ions) impinging on 16 the tokamak walls was analyzed, including ion energy distribution.

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21 1. Introduction

The interactions of neutral and charged hydrogen particles with plasma-facing materials is of prime importance in the present-day fusion devices and their models, as they define D/T fuel recycling and inventory [1,2]. For that reason, measuring of the hydrogen flux from plasma, of hydrogen particle energetic distribution, and of the molecular composition and ion charge, takes on a great interest.

Hydrogen-wall interactions are governed to a great degree by the state of wall surface which conditions the processes of hydrogen absorption/desorption/accumulation. The probability, α , of the absorption of low-energy hydrogen particles (~1 to hundreds eV) in metals through a nonmetallic film typically covering the metal surface radically depends on the film thickness. In the case of a monolayer

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film, this probability is close to that for a clean metallic sur-36 face. However, it may decrease dramatically, if the film 37 thickness exceeds one monolayer. In reality, the thickness 38 of nonmetallic coating depends on temperature: at high 39 enough temperatures, typically only one nonmetallic mono-40 layer exists at the surface under vacuum conditions, and 41 hence the probability of absorption of hydrogen particles 42 whose kinetic, internal or chemical energy exceeds 1 eV 43 (suprathermal hydrogen) is very high. At lower tempera-44 tures, both monolayer and polyatomic coatings are possi-45 ble, and, correspondingly, α may vary over a wide range. 46

The present work was undertaken to investigate the role 47 played by the nonmetallic coatings on plasma-facing materials in D/T inventory and in the recycling of low energy 49 hydrogen particles in tokamak environments; another 50 goal is developing of a method of suprathermal particle 51 diagnostics. 52

A resistively heated absorption probe of Group V met-33 als (niobium and vanadium) is used to trap the flux of hydrogen particles, the amount of which is measured dur-55

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M.E. Notkin et al. | Nucl. Instr. and Meth. in Phys. Res. B xxx (2006) xxx-xxx

ing the post-exposure thermal desorption of the absorbedhydrogen.

58 The choice of the probe material is due to a very favor-59 able combination of such parameters of hydrogen interac-60 tion with the Group V metals as a large positive heat of 61 hydrogen solution, a high mobility of solute hydrogen 62 depending only weakly on metal temperature, and a high 63 chemical activity of the metal surface giving rise to stable 64 surface films of nonmetallic elements effectively impeding 65 the thermal desorption of solute hydrogen. Due to that, probes of these metals can operate over a wide range of 66 67 tokamak operation parameters, including high particle flux densities, short pulse duration, and also in devices having a 68 69 high level of the background molecular hydrogen where 70 other known diagnostic methods may be inapplicable [2,3].

71 The CASTOR tokamak [4] offers a good opportunity for 72 staging of such experiments due to a relatively high, up to 10^{20} H/(m² s), flux of suprathermal hydrogen particles, an 73 74 easy change of probe samples, and the vacuum system free 75 of carbon-containing impurities. On the other hand, one has to note a relatively high, up to 3×10^{-2} Pa, level of 76 the background hydrogen, and a short, <35 ms, duration 77 78 of the plasma pulse.

79 2. Experimental

Schematic layout of the experimental setup is presented in Fig. 1. A movable resistively heated absorption probe (AP) of the investigated material (0.02 mm niobium foil, or 0.025 vanadium foil) with plasma facing surface area, S_{probe} , ~50 cm² is placed in a bakeable vacuum chamber connected to a CASTOR tokamak port through a gate valve. The AP is fixed at movable rod of magnetic drive,



Fig. 1. Experimental layout.

87 and it can be transported along the system axis over a \sim 100 cm distance. AP can be located at different distances 88 from the tokamak plasma during the absorption experi-89 ments, and it can be moved into the experimental chamber 90 91 which can be separated from the torus during the Castor bake-up, as well as for AP conditioning, for the analytical 92 93 degassing experiments, experiments on AP surface modification, etc. 94

95 The AP chamber is pumped by a 200 l/s TMP (40 l/s for 96 hydrogen). The pressure is measured by a mass-spectrometer and a set of vacuum gauges. The state of AP surface 97 (coating type and thickness) can be varied in situ by con-98 trolled reactions with specially introduced chemically 99 active gases. One can obtain clean Nb(V) surface by heat-100 ing the probe to a high temperature. At the first step of this 101 study, the probes were only heated to moderate tempera-102 tures (1900 K for Nb, and 1600 K for V) in vacuum, which 103 leads to the dissolution of polyatomic oxide films, with 104 only O monolayer remaining on the surface [5]. 105

Measurements of the absorption of hydrogen from 106 plasma were performed in two steps. First, the probe was 107 moved into the CASTOR torus and put at a desirable 108 distance from the plasma. Then, an operational hydrogen 109 110 pressure was established, and a plasma shot was executed. After the exposure to plasma, AP was moved back into the 111 separate chamber. The amount of the absorbed hydrogen 112 was measured at H₂ thermal desorption from the probe 113 by its heating to 1300 K with the gate valve closed. A 114 typical example of thermodesorption curve is presented 115 in Fig. 2, curve 1. The area under this curve is propor-116 tional to the number of hydrogen particles absorbed 117 both from the plasma and from the background molecular 118 119 hydrogen.

Second, the above procedure was fully repeated, but 120 without plasma shot. A corresponding thermodesorption 121 curve is presented in Fig. 2, curve 2; it refers to the desorp-122 tion of hydrogen absorbed only from the background 123 hydrogen. The number of suprathermal hydrogen particles 124 absorbed from the plasma is proportional to the difference 125 of the areas under the curves 1 and 2 (the hatched area in 126 Fig. 2). 127



Fig. 2. Hydrogen thermodesorption from the probe at the permanent pumping with a speed 40 l/s for hydrogen: (1) suprathermal particles and (2) molecular hydrogen.

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M.E. Notkin et al. | Nucl. Instr. and Meth. in Phys. Res. B xxx (2006) xxx-xxx

128 3. Experimental results and discussion

129 3.1. Dependence of the absorption of hydrogen from

130 plasma on probe location

131 The dependence of the amount of hydrogen accumu-132 lated from the suprathermal hydrogen flux on probe loca-133 tion is presented in Fig. 3; it demonstrates a strong 134 dependence of hydrogen accumulation on the distance between AP and plasma. AP position just in line with the 135 tokamak wall is indicated in Fig. 3 as "0" position, which 136 corresponds to a 94 mm distance from the tokamak vessel 137 centre. The "minus" sign refers to an AP shift toward the 138 139 plasma.

140 The flux of suprathermal particles consists of both neu-141 tral atoms and ions. As the concentration of neutral hydrogen atoms is presumably equidistributed over the whole 142 143 range of possible AP location, the observed dependence 144 is most probably due to the ions whose flux is different at different distances from plasma because of shielding by 145 146 the CASTOR limiter, walls, etc., and due to the presence 147 of magnetic field. One can assume that the independence



Fig. 3. Dependence of hydrogen accumulation on probe location. T(exposure) = 373 K, T(desorption) = 1123 K, floating potential.



Fig. 4. Temperature dependence of hydrogen molecule absorption probability.

of hydrogen accumulation on probe location (the right 148 branch of the curve in Fig. 3) indicates at the absence of 149 an input from ions. Based on this result, the experiments 150 with neutral particles were performed over this range of 151 probe location. 152

3.2. Absorption of hydrogen molecules 153

One can characterize the state of AP surface with regard 154 to hydrogen absorption by the dependence of the kinetics 155 of molecular hydrogen absorption on probe temperature 156 (Fig. 4). We defined the sticking coefficient of hydrogen 157 molecules, α_{H_2} , from our experimental data as follows 158 $\alpha_{\rm H_2} = N(\rm abs)/N(\rm inc)$. Here N(inc), the amount of hydro-159 gen molecules falling onto the probe surface, equals $K_1 \times$ 160 $P_1(H_2) \times t$ where $P_1(H_2)$ is hydrogen pressure measured 161 during AP exposure, t is exposure time and K_1 is a propor-162 tionality coefficient. The amount of hydrogen molecules 163 absorbed by the probe, N(abs), equals to the amount of 164 hydrogen desorbed from AP at thermodesorption, and it 165 can be experimentally measured as described in Section 2. 166

As it follows from the data presented in Fig. 4, $\alpha_{\rm H_2} =$ 167 $2.1 \times 10^{-2} \exp{-(28.2 \text{ kJ/kT})}$ for the vanadium probe, 168 and $\alpha_{H_2} = 1.5 \times 10^{-3} \exp{-(13.6 \text{ kJ/kT})}$ for the niobium 169 probe. These values are typical of an H₂ molecule absorp-170 tion by Nb and by V through a surface covered by nonme-171 tallic (e.g. O) impurity monolayer resulting from the 172 specimen annealing under vacuum conditions [5] (note that 173 $\alpha_{\rm H_2} = 0.1-0.3$ for a clean Nb surface [6]). According to our 174 experience, that means that the probe surfaces are in a state 175 giving rise to superpermeation, at least over the investi-176 gated temperature range, and hence one can reckon on 177 an effective absorption of suprathermal particles. 178

3.3. Absorption of hydrogen atoms 179

The number of suprathermal particles absorbed during 180 the tokamak discharge, $N_{\rm H}$, was measured as function of 181 AP temperature (Fig. 5, curve 3) and of plasma pulse 182 length (Fig. 6). One can see that the amount of absorbed 183



Fig. 5. Temperature dependencies of hydrogen atom absorption probability, (1) vanadium, (2) niobium and of hydrogen accumulation during plasma discharge (3).

M.E. Notkin et al. | Nucl. Instr. and Meth. in Phys. Res. B xxx (2006) xxx-xxx



Fig. 6. Dependence of hydrogen absorption on pulse length.

suprathermal particles depends weakly on probe temperature and is approximately proportional to pulse length. The number of hydrogen atoms absorbed during plasma discharge amounts up to $\sim 10^{17}$ atom H, depending on probe temperature and on plasma pulse length (Figs. 5 and 6).

190 Concerning the probability of hydrogen atom absorp-191 tion, we take its magnitude from the known data on the 192 implantation coefficient, $\alpha_{\rm H}$, at sufficiently high tempera-193 tures: when it is guaranteed that the oxygen coating does 194 not exceed one monolayer. For instance, one can take 195 $\alpha_{\rm H}\approx 0.2$ as a typical value for the energy range around 196 10 eV ($\alpha_{\rm H} \approx 0.2$ is ascribed to our experimental data for suprathermal hydrogen: Fig. 5 curves 1, 2, left axis). As 197 198 one could have expected it, the probabilities of hydrogen 199 atom absorption for niobium (curve 2) and vanadium 200 (curve 1) probes are very close to one the other.

201 3.4. Bias experiment

The study of the absorption of hydrogen ions was carried out with the same procedure as above (Section 2), but vanadium probe was electrically biased with respect to the CASTOR walls and was placed at a position where there is a measurable ion flux (see above). A typical dependence of the ion current on probe bias voltage is presented in Fig. 7 for two AP positions (see also Fig. 3).



Fig. 7. Dependence of ion current onto the probe on probe bias voltage for two radial positions of the probe: (1) $r_p = 94 \text{ mm}$ ("0" position), (2) $r_p = 84 \text{ mm}$.



Fig. 8. Dependence of hydrogen accumulation on ion energy.

The results of the corresponding thermodesorption 209 experiments (Fig. 8) demonstrate that hydrogen absorption 210 does not depend on probe bias over the range +(40-100) V. 211 One can suggest that such an independence means that the 212 input of ions into hydrogen absorption is negligible over 213 this energy range, hydrogen accumulation in the probe 214 being due only to the input of neutrals. 215

The absorption of neutral hydrogen atoms does not 216 depend on probe bias and thus it was possible to measure 217 it separately. The results of these measurements are pre-218 sented in Fig. 8 as a gray bar. The difference between the 219 total desorption from the probe and the desorption indicat-220 ing the input of neutral atoms yields the input of hydrogen 221 222 ions. It follows from these data that the probability of hydrogen ion absorption weakly depends on ion energy 223 over the whole investigated energy range and equals 224 $\alpha_{\rm H} = 0.75 \pm 0.05$. These numeric estimates are made on 225 the presumption that the ion flux consists of H^+ ions [4]. 226

Besides, AP being actually a Langmuir probe, its biasing 227 permits to measure energy distribution of ions in the vicinity of CASTOR wall. 229

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4. Conclusion

231 From a practical point of view, the possibility of a reliable registration of suprathermal hydrogen particles (atoms 232 and ions) from tokamak plasma with the absorption probes 233 made of superpermeable niobium and vanadium was dem-234 onstrated in the CASTOR tokamak conditions with its 235 short plasma pulse and a relatively high background H_2 236 pressure. The absorption of suprathermal hydrogen parti-237 cles from plasma was investigated as function of probe dis-238 239 tance from the plasma, probe bias, and plasma discharge duration. These results provide the opportunity to study 240composition of the hydrogen flux (molecules, atoms, ions) 241 impinging upon the CASTOR liner and to evaluate the 242 fluxes of each component, as well as the energy distribution 243 of hydrogen ions. 244

Absorption of 7 eV Franck–Condon atoms [4] and of 245 low energy (up to hundreds of eV) ions by the Group V metals, niobium and vanadium, with the surface covered with 247 oxygen monolayer film (Section 2) was studied from the 248 viewpoint of physics. It was demonstrated that such a film 249

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M.E. Notkin et al. | Nucl. Instr. and Meth. in Phys. Res. B xxx (2006) xxx-xxx

does not impede the absorption of suprathermal hydrogen 250 251 particles, but it suppresses their recycling, which results in an inventory of up to $\sim 10^{17}$ atom H from the plasma, sub-252 stantially exceeding that from hydrogen molecules. 253

The absorption of suprathermal hydrogen particles by 254 255 niobium and vanadium probes was found to be very close 256 to each other. The absorption probability of hydrogen 257 atoms was found to weakly depend on metal temperature 258 over temperature range 300-1000 K. The absorption probability of hydrogen ions does not depend on ion energy 259 260 over the range 20-100 eV.

261 The measured number of H atoms absorbed during a standard 25 ms discharge is in a reasonable agreement with 262 that estimated from the plasma parameters. 263

264 The continuation of the present work will be aimed at 265 modifications of the probe surface with the purpose to strongly suppress the absorption of hydrogen molecules 266 267 and to increase in this way the sensitivity of suprathermal particle registration. Another promising idea is developing 268 269 of a method to measure the amount of absorbed hydrogen based on the dependence of probe material resistivity on 270

271 the concentration of solute hydrogen.

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